

Phytoremediation potential of native plants grown in the vicinity of Ahangaran lead–zinc mine (Hamedan, Iran)

J. Nouri · B. Lorestani · N. Yousefi · N. Khorasani ·
A. H. Hasani · F. Seif · M. Cheraghi

Received: 23 November 2009 / Accepted: 8 April 2010 / Published online: 8 May 2010
© Springer-Verlag 2010

Abstract This study aims to assess the extent of metal accumulation by plants found in a mining area in Hamedan Province in the central west part of Iran. It also investigates to find suitable plants for phytoextraction and phytostabilization as two phytoremediation strategies. Plants with a high bioconcentration factor (BCF) and low translocation factor (TF) have the potential for phytostabilization while plants with both BCFs and TFs greater than one have the potential to be used for phytoextraction. In this study, shoots and roots of the 12 plant species and the associated soil samples were collected. The collected samples were then analyzed by measurement of total concentrations of trace elements (Pb, Zn, Mn and Fe) using atomic absorption spectrophotometer. Simultaneously, BCF and TF parameters were calculated for each element. Results showed that although samples suitable for phytoextraction of Pb, Zn, Mn and Fe and phytostabilization of Fe were not detected, *Scrophularia scoparia* was the most suitable for phytostabilization of Pb, *Centaurea virgata*, *Echinophora platyloba* and *Scariola orientalis* had the potential for phytostabilization of Zn and *Centaurea virgata* and *Cirsium congestum* were the most efficient in phytostabilization of Mn. Present study showed that native plant species growing on contaminated sites may have the potential for phytoremediation.

Keywords Phytoremediation · Lead–zinc mine · Heavy metals · Native plants · Translocation factor · Bioconcentration factor

Introduction

Waste released from mines causes major environmental problems all over the world. Heavy metals content in mine waste are much higher than in contaminated soils. They are harmful to humans and animals and tend to accumulate in the food chain (McDowell 1992; Nouri et al. 2008). Activities such as mining and smelting of metal ores, industrial emissions and applications of insecticides and fertilizers have all contributed to elevate levels of heavy metals in the environment (Alloway 1994; Atafar et al. 2008). Several technologies are available to remediate soils contaminated by heavy metals. However, many of these technologies are costly, e.g., excavation of contaminated material and chemical/physical treatment or do not achieve a long-term nor esthetic solution (Cao et al. 2002; Mulligan et al. 2001). Metal accumulation by plants is affected by many factors. In general, variations in plants species, the growth stage of the plants and element characteristics control absorption, accumulation and translocation of metals (Guilizzoni 1991; Samarghandi et al. 2007; Behbahaninia et al. 2009). Phytoremediation can provide a cost-effective, long-lasting and esthetic solution for remediation of contaminated sites (Ma et al. 2001). Phytoremediation is the process of application of green plants to remove or render environmental contaminants harmless (Cunningham and Berti 1993; Nouri et al. 2009). There is an evidence that plants such as *Sebera acuminata* and *Thlaspi caerulescens* can accumulate heavy metals in their tissues (Cunningham and Ow 1996), *Arabidopsis thaliana*

J. Nouri · A. H. Hasani
Graduate School of the Environment and Energy, Science
and Research Branch, Islamic Azad University, Iran

B. Lorestani (✉) · N. Yousefi · F. Seif · M. Cheraghi
Islamic Azad University, Hamedan Branch, Hamedan, Iran
e-mail: lorestani_B@iauh.ac.ir

N. Khorasani
Department of Environmental Science, College of Natural
Resources, University of Tehran, Tehran, Iran

(Delhaize 1996), *Typha latifolia* and *Phragmites australis* (Ye et al. 2001). *T. latifolia* and *P. australis* have been successfully used for phytoremediation of Pb/Zn mine (Ye et al. 1997a, b). Remediation of heavy metals by plant species can be divided into three groups: phytoextraction, metal accumulating plants are planted on contaminated soil and later harvested in order to remove metals from the soil; rhizofiltration, roots of metal accumulating plants absorb metals from polluted effluents and are later harvested to diminish the metals in the effluent; and phytostabilisation, metal-tolerant plants are used to reduce the mobility of metals, thus, the metals are stabilized in the substrate (Salt et al. 1995; Abdel-Ghani et al. 2007). Plants with both bioconcentration factors and translocation factors greater than one (TF and $BCF > 1$) have the potential to be used in phytoextraction. Besides, plants with bioconcentration factor greater than one and translocation factor less than one ($BCF > 1$ and $TF < 1$) have the potential for phytostabilization (Yoon et al. 2006). There has been a continuing interest in searching for native plants that are tolerant towards heavy metals (Shu et al. 2002 and McGrath and Zhao 2003).

The overall objectives of this research were: (1) to determine the concentrations of Pb, Zn, Mn and Fe in plant biomass growing on a contaminated site; (2) to compare metal concentrations in the aboveground biomass with those in roots and in soils, and (3) to assess the feasibility of using of these plants for phytoremediation purpose. Results of this study should provide insight for using native plants to remediate metal-contaminated sites.

Materials and methods

Site characterization

The plant and soil samples used in this study were collected from a known metal-contaminated site located in an urban area, called Ahangaran lead–zinc mine of Malayer city, southwest of Hamedan (Fig. 1). The site has been vacant, occupied approximately 1,600,000 m², and is covered mainly by grasses. Human activities such as mining have contributed to metal concentrations in this site. Contamination of heavy metals was mainly concentrated in the top 20 cm at the site.

Selected characteristics of the soil samples collected from this study are shown in Table 1.

Plant sampling and analysis

Twelve plant species and the associated soil samples were collected in the surrounding area of Ahangaran mine from

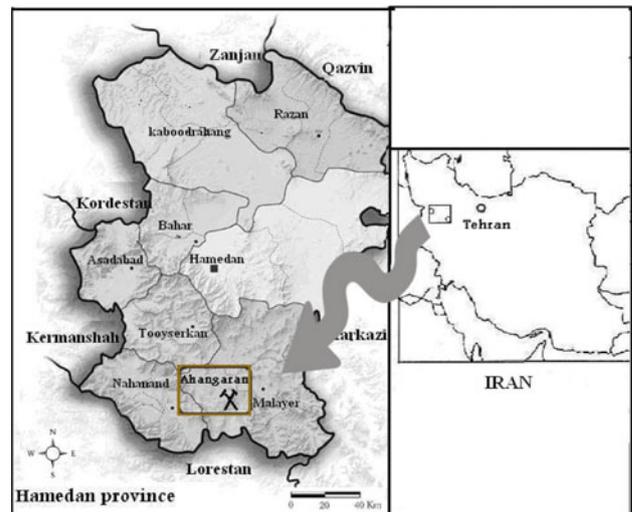


Fig. 1 Location of the study area

May to June 2007. The studied species consisted of 11 genera and 8 families (Table 2), of which 4 species belonged to Asteraceae, forming the most dominant component in metal-polluted sites.

Plant samples were thoroughly washed with running tap water and rinsed with deionized water to remove any soil particles attached to the plant surfaces, then oven dried (70°C) to constant weight. The dried tissues were weighed and ground into powder for metal concentration analysis. Metal contents (Fe, Zn, Pb and Mn) of the plant samples were extracted by acid digestion followed by measurement of total concentrations of all elements of interest using atomic absorption spectrophotometer (GBC Avanta, Australia) (Ward et al. 1975).

Soil sampling and analysis

Soils were sampled from the same sites and location points as the plants. The top 20 cm soil from between the plant roots was collected, air-dried for 3 weeks, and then sieved through a 2-mm mesh. Samples were then analyzed for total metals (Fe, Zn, Pb and Mn). Total metal contents were extracted by acid digestion. Metal contents were measured by atomic absorption spectrophotometer (GBC Avanta, Australia), (Sposito 1989). Soil samples were also analyzed for pH, electrical conductivity (EC) and total P. The pH and EC values (solid:distilled water = 1:5) of soil samples were measured by a pH meter and an EC meter, respectively (Thomas 1996; Roades 1996). Total phosphorus was measured using the method of Olsen (Rowell 1994) and soil texture, was determined according to Tan (1995).

Table 1 Soil characteristics of the surrounding area of Ahangaran mine (mean ± SE)

Parameters	EC (µs/m)	pH	P (mg/kg)	CaCO ₃ (mg/kg)	Clay (%)	Silt (%)	Sand (%)
Value	385 ± 14	7.8 ± 0.02	56 ± 12	18 ± 3.7	9 ± 0.5	20 ± 2.6	71 ± 7.1

Table 2 Species composition in surrounding area of Ahangaran mine

Species no.	Scientific name	Family
1	<i>Euphorbia macroclada</i> Boiss.	Euphorbiaceae
2	<i>Centaurea virgata</i> Lam.	Asteraceae
3	<i>Reseda lutea</i> L.	Resedaceae
4	<i>Echinophora platyloba</i> DC.	Geramineae
5	<i>Cichorium intybus</i> L.	Asteraceae
6	<i>Tamarix ramosissima</i> Ledeb.	Tamaricaceae
7	<i>Cirsium congestum</i> Fisch. and C. A.	Asteraceae
8	<i>Scrophularia scoparia</i> Pennell	Scrophulariaceae
9	<i>Chaerophyllum macropodium</i> Boiss.	Umbelliferae
10	<i>Cardaria draba</i> (L.) Desv.	Cruciferae
11	<i>Scariola orientalis</i> (Boiss.) Sojak.	Asteraceae
12	<i>Reseda alba</i> L.	Resedaceae

Table 3 Pb concentrations in soil and plant samples (mg/kg)

Scientific name	Roots	Shoots	Soil
<i>Euphorbia macroclada</i> Boiss.	3,809	8,095	9,451
<i>Centaurea virgata</i> Lam.	1,180	1,927	8,446
<i>Reseda lutea</i> L.	98	1,774	7,367
<i>Echinophora platyloba</i> DC.	1,421	10,121	10,426
<i>Cichorium intybus</i> L.	ND	15	12,140
<i>Tamarix ramosissima</i> Ledeb.	130	2,010	10,401
<i>Cirsium congestum</i> Fisch. and C. A.	6,750	2,934	7,389
<i>Scrophularia scoparia</i> Pennell	577	6,270	6,847
<i>Chaerophyllum macropodium</i> Boiss.	3,421	149	6,895
<i>Cardaria draba</i> (L.) Desv.	676	3,171	3,847
<i>Scariola orientalis</i> (Boiss.) Sojak.	1,204	9,017	9,140
<i>Reseda alba</i> L.	1,743	703	9,535

ND not detected

Calculation of TF and BCF

The ability of plants to tolerate and accumulate heavy metals is useful for phytoextraction and phytostabilization purpose measured using translocation factor (TF) and bioconcentration factor (BCF) defined as the ratio of metal concentration in plant shoots to roots ($[Metal]_{Shoot}/[Metal]_{Root}$) and the ratio of metal concentration in plant roots to soils ($[Metal]_{Root}/[Metal]_{Soil}$), respectively (Yoon et al. 2006).

Results and discussion

Soil characteristics and metal concentrations

As shown in Table 1, soils in the surrounding area of mine were slightly alkaline; with an average pH of approximately 7.8. The pH conditions were suitable for plant growth. The average of EC, total P concentration and CaCO₃ were 385 µs/m, 56 mg/kg and 18 mg/kg, respectively. Also, texture of soil was loam-sandy in the soil collected from the surrounding area of mine.

The present study shows that some plants can colonize sites with a wide range of metal concentrations in the soils. According to Kabata-Pendias and Pendias (1984), 2–200 mg/kg Pb, 70–400 mg/kg Zn and 300–500 mg/kg Mn and insignificant data for Fe would be considered toxic to plants based on total fractions in soil. The metal (Fe, Zn, Pb and Mn) contents in the surrounding area of mine

greatly exceed these ranges (Tables 3, 4, 5, 6); therefore, the 12 plant species grown in these contaminated sites have exhibited high metal tolerance.

Metal concentrations in plants

Metal concentrations in plants vary with plant species (Alloway et al. 1990). Plant uptake of heavy metals from soil occurs either passively with the mass flow of water into the roots, or through active transport crosses of the plasma membrane of root epidermal cells. Under normal growing conditions, plants can potentially accumulate certain metal ions an order of magnitude greater than the surrounding medium (Kim et al. 2003). Concentrations of Pb, Mn, Zn and Fe in collected plant species are provided in Tables 3, 4, 5, 6. Toxic concentrations of heavy metals for various plant species are 10–100, 400–1,000, 100–400 and 80–1,450 mg/kg for Pb, Mn, Zn and Fe, respectively (Kabata-Pendias and Pendias 1984). According to Tables 3, 4, 5, 6, in the most (62.5%) of the plant samples, heavy metal contents were higher than toxic levels. Total Pb concentrations in the plant roots ranged from non-detectable to as high as 6,750 mg/kg and plant shoots from 15 to as high as 9,017 mg/kg, with the maximum level in the roots of *Cirsium congestum* and shoots of *Scariola orientalis*. Total Zn concentrations in the plant roots ranged from non-detectable to 2,938 mg/kg and plant shoots from non-detectable to 3,538 mg/kg, with the maximum level in the roots of *Reseda lutea* and shoots of *Scariola orientalis*. Total concentrations of Mn in the plant roots ranged from

Table 4 Zn concentrations in soil and plant samples (mg/kg)

Scientific name	Roots	Shoots	Soil
<i>Euphorbia macroclada</i> Boiss.	81	967	1,645
<i>Centaurea virgata</i> Lam.	762	3,155	1,825
<i>Reseda lutea</i> L.	2,938	313	2,593
<i>Echinophora platyloba</i> DC.	1,925	2,150	2,016
<i>Cichorium intybus</i> L.	ND	32	1,389
<i>Tamarix ramosissima</i> Ledeb.	351	2,624	3,967
<i>Cirsium congestum</i> Fisch. and C. A.	1,751	590	1,916
<i>Scrophularia scoparia</i> Pennell	341	68	3,638
<i>Chaerophyllum macropodium</i> Boiss.	1,425	356	4,205
<i>Cardaria draba</i> (L.) Desv.	1,757	860	3,909
<i>Scariola orientalis</i> (Boiss.) Sojak.	1,429	3,538	2,119
<i>Reseda alba</i> L.	73	ND	2,845

ND not detected

Table 5 Mn concentrations in soil and plant samples (mg/kg)

Scientific name	Roots	Shoots	Soil
<i>Euphorbia macroclada</i> Boiss.	57	149	843
<i>Centaurea virgata</i> Lam.	156	1,220	963
<i>Reseda lutea</i> L.	29	116	975
<i>Echinophora platyloba</i> DC.	1,283	303	1,361
<i>Cichorium intybus</i> L.	235	62	962
<i>Tamarix ramosissima</i> Ledeb.	76	15	909
<i>Cirsium congestum</i> Fisch. and C. A.	459	538	512
<i>Scrophularia scoparia</i> Pennell	36	386	751
<i>Chaerophyllum macropodium</i> Boiss.	342	177	596
<i>Cardaria draba</i> (L.) Desv.	128	261	609
<i>Scariola orientalis</i> (Boiss.) Sojak.	105	626	751
<i>Reseda alba</i> L.	325	17	695

ND not detected

Table 6 Fe concentrations in soil and plant samples (mg/kg)

Scientific name	Roots	Shoots	Soil
<i>Euphorbia macroclada</i> Boiss.	57	149	843
<i>Centaurea virgata</i> Lam.	156	1,220	963
<i>Reseda lutea</i> L.	1,611	3,617	12,433
<i>Echinophora platyloba</i> DC.	4,045	10,764	12,168
<i>Cichorium intybus</i> L.	2,752	10,819	13,889
<i>Tamarix ramosissima</i> Ledeb.	7,808	4,311	11,881
<i>Cirsium congestum</i> Fisch. and C. A.	53	117	10,315
<i>Scrophularia scoparia</i> Pennell	ND	215	9,198
<i>Chaerophyllum macropodium</i> Boiss.	3,652	9,709	10,617
<i>Cardaria draba</i> (L.) Desv.	1,306	2,557	10,515
<i>Scariola orientalis</i> (Boiss.) Sojak.	1,645	1,370	9,201
<i>Reseda alba</i> L.	1,637	1,938	9,574

ND not detected

29 to as high as 1,283 mg/kg and plant shoots from 15 to as high as 1,220 mg/kg, with the maximum being in the roots of *Echinophora platyloba* and shoots of *Centaurea virgata*. Fe concentrations in plant roots differed among species at the polluted site from non-detectable to 7,808 mg/kg and in shoots from 117 to 10,819 mg/kg, with the maximum content in the roots of *Tamarix ramosissima* and shoots of *Cichorium intybus*.

Accumulation and translocation of metals in plants

The plant's ability to accumulate metals from soils and to translocate metals from the roots to the shoots can be estimated using the BCF and plant's ability is measured using the TF, respectively. Enrichment occurs when a contaminant taken up by a plant is not degraded rapidly, resulting in an accumulation in the plant. The process of phytoextraction generally requires the translocation of heavy metals to the easily harvestable plant parts, i.e., shoots (Yoon et al. 2006). Plant species which have strong ability to reduce metal translocation from roots to shoots are suitable as phytostabilizers for metal-contaminated soils such as mine tailing (Deng et al. 2004). By comparing BCF and TF, the ability of different plants in taking up metals from soils and translocating them to the shoots can be compared. Plants exhibiting TF and particularly BCF values greater than one are suitable for phytoextraction and plants with BCF greater than one and TF less than one have the potential for phytostabilization (Yoon et al. 2006).

As shown in Table 7, among the sampled plants, none of them were suitable for phytoextraction of Pb, Zn, Mn and Fe, and phytostabilization of Fe but *Scrophularia scoparia* was most suitable for phytostabilization of Pb (BCF = 1.43 and TF = 0.09), *Centaurea virgata*, *Echinophora platyloba* and *Scariola orientalis* had the potential for phyto-stabilization of Zn (BCF = 1.73, 1.07 and 1.67, and TF = 0.24, 0.89 and 0.40, respectively) and *Centaurea virgata* and *Cirsium congestum* were the most efficient in phyto-stabilization of Mn (BCF = 1.27, 1.07 And TF = 0.13 and 0.85, respectively).

Conclusion

This study was conducted to screen plants growing on a contaminated site to determine their potential for metal accumulation. Based on among the 12 sampled plant species, metal translocation into shoots appears to be very restricted in the most plant species so that harvesting plants will not be an effective source of metal removal in this site. However, in the view of toxicology, this could be a desirable property, as metals would not pass into the food chain via herbivores, and thus avoid potential risk to the

Table 7 Accumulation and translocation of Pb, Zn, Mn and Fe in the selected plants

Scientific name	Bioconcentration factor (BCF)				Translocation factor (TF)			
	Pb	Zn	Mn	Fe	Pb	Zn	Mn	Fe
<i>Euphorbia macroclada</i> Boiss.	0.86	0.59	–	0.30	0.47	0.08	0.38	0.44
<i>Centaurea virgata</i> Lam.	0.23	1.73	1.27	0.88	0.61	0.24	0.13	0.37
<i>Reseda lutea</i> L.	0.24	0.12	0.12	0.78	0.05	9.39	0.25	0.25
<i>Echinophora platyloba</i> DC.	0.97	1.07	0.22	0.36	0.14	0.89	4.20	1.81
<i>Cichorium intybus</i> L.	0.00	0.02	0.06	0.01	–	–	3.80	0.45
<i>Tamarix ramosissima</i> Ledeb.	0.19	0.67	0.02	0.02	0.06	0.13	5.06	–
<i>Cirsium congestum</i> Fisch. and C. A.	0.40	0.31	1.05	0.91	0.23	2.97	0.85	0.38
<i>Scrophularia scoparia</i> Pennell	1.43	0.02	0.51	0.24	0.09	5.01	0.09	0.51
<i>Chaerophyllum macropodium</i> Boiss.	0.02	0.08	0.30	0.15	22.96	4.00	1.93	1.20
<i>Cardaria draba</i> (L.) Desv.	0.93	0.22	0.43	0.20	0.21	2.04	0.49	0.84
<i>Scariola orientalis</i> (Boiss.) Sojak.	0.99	1.67	0.86	0.23	0.13	0.40	0.16	1.11
<i>Reseda alba</i> L.	0.07	–	0.02	0.02	2.48	–	19.12	15.76

BCF metal concentration ratio of plant roots to soil, TF metal concentration ratio of plant shoots to roots

environment. Therefore, plants with low shoot accumulation should be used in order to stabilize the metals and reduce the metal dispersion through grazing animals or at leaf senescence. In general, sampled plants are almost exclusively annual and perennial herbaceous species and produce relatively large biomass; therefore, their utility for phytoremediation is possible.

Acknowledgments The authors would like to express their appreciation to University of Hamedan and Graduate School of the Environment and Energy for their facilities and kind support.

References

Abdel-Ghani NT, Hefny M, El-Chagbaby GAF (2007) Removal of lead from aqueous solution using low cost abundantly available adsorbents. *Int. J. Environ. Sci. Tech* 4(1):67–73

Alloway BJ (1994) Toxic metals in soil–plant systems. John Wiley and Sons, Chichester, UK

Alloway BJ, Jackson AP, Morgan H (1990) The accumulation of cadmium by vegetables grown on soils contaminated from a variety of sources. *Sci Total Environ* 91:223–236

Atafar Z, Mesdaghinia A, Nouri J, Homae M, Yunesian M, Ahmadi Moghaddam M, Mahvi AH (2008) Effect of fertilizer application on soil heavy metal concentration. *Environ. Monitor. Assess.* 1–7 (in press)

Behbahani A, Mirbagheri SA, Khorasani N, Nouri J, Javid AH (2009). Heavy metal contamination of municipal effluent in soil and plants. *J. Food, Agri. Environ.* 7(3–4):851–856

Cao X, Ma LQ, Chen M (2002) Impacts of phosphate amendments on lead biogeochemistry in a contaminated site. *Environ Sci Technol* 36:5296–5304

Cunningham SD, Berti WR (1993) Remediation of contaminated soils with green plants: an overview. *In Vitro Cell Dev Biol* 29:207–212

Cunningham SD, Ow DW (1996) Promises of phytoremediation. *Plant Physiol* 110:15–719

Delhaize A (1996) A metal accumulator mutant of *Arabidopsis thaliana*. *Plant Physiol* 111:849–855

Deng H, Ye ZH, Wong MH (2004) Accumulation of lead, zinc, copper and cadmium by 12 wetland plant species thriving in metal-contaminated sites in China. *Environ Pollut* 132:29–40

Guilizzoni P (1991) The role of heavy metals and toxic materials in the physiological ecology of submerged macrophytes. *Aquat Bot* 41:87–109

Kabata-Pendias A, Pendias H (1984) Trace elements in soils and plants. CRC Press, Florida

Kim IS, Kang KH, Johnson-Green P, Lee EJ (2003) Investigation of heavy metal accumulation in *Polygonum thunbergii* for phytoextraction. *Environ Pollut* 126:235–243

Ma LQ, Komar KM, Tu C, Zhang WA (2001) A fern that hyperaccumulates arsenic. *Nature* 409:579

McDowell LR (1992) Minerals in animal and human nutrition. Academic Press, San Diego

Mcgrath SP, Zhao FJ (2003) Phytoextraction of metals and metalloids from contaminated soils. *Curr Opin Biotechnol* 14:1–6

Mulligan CN, Yong RN, Gibbs BF (2001) Remediation technologies for metal-contaminated soils and groundwater, an evaluation. *Eng Geol* 60:193–207

Nouri J, Mahvi AH, Jahed GR, Babaei AA (2008) Regional distribution pattern of groundwater heavy metals resulting from agricultural activities. *Environ. Geol* 55(6):1337–1343

Nouri J, Khorasani N, Lorestani B, Karami M, Hassani AH, Yousefi N (2009) Accumulation of heavy metals in soil and uptake by plant species with phytoremediation potential. *Environ Earth Sci*, pp 1–9. doi:10.1007/s12665-009-0028-2 (in press)

Roades JD (1996) Salinity: electrical conductivity and total dissolved solids methods of soil analysis, chemical methods. American Society of Agronomy, Madison, WI

Rowell DL (1994) Soil science: methods and applications. Longman, Harlow

Salt DE, Blaylock M, Kumar NPBA, Dushenkov V, Ensley BD, Chet I, Raskin I (1995) Phytoremediation: a novel strategy for the removal of toxic metals from the environment using plants. *Biotechnology* 13:468–474

Samarghandi MR, Nouri J, Mesdaghinia AR, Mahvi AH, Vaezie F, Nasserli S (2007) Efficiency removal of phenol, lead and cadmium

- by means of UV/TiO₂/H₂O₂ processes. *Int. J. Environ. Sci. Tech* 4(1):19–25
- Shu WS, Ye ZH, Lan CY, Zhang ZQ, Wong MH (2002) Lead, zinc and copper accumulation and tolerance in populations of *Paspalum distichum* and *Cynodon dactylon*. *Environ Pollut* 120:445–453
- Sposito G (1989) *The chemistry of soils*. Oxford University Press, New York
- Tan KH (1995) *Environmental soil science*. Marcel Dekker, Inc., New York
- Thomas GW (1996) Soil pH and soil acidity. In: Klute A (ed) *Methods of soil analysis. Part 3. Chemical methods*. American Society of Agronomy, Madison, WI
- Ward NI, Reeves RD, Brooks RR (1975) Lead in soil and vegetation. *Environ Pollut* 9:243–251
- Ye ZH, Baker AJM, Wong MH, Willis AJ (1997a) Copper and nickel uptake, accumulation and tolerance in populations of *Typha latifolia* L. *New Phytol* 136:469–480
- Ye ZH, Baker AJM, Wong MH, Willis AJ (1997b) Zinc, lead and cadmium tolerance, uptake and accumulation by the common reed, *Phragmites australis* (Cav.) Trin Ex Steudel. *Ann Bot* 80:363–370
- Ye ZH, Whiting SN, Lin ZQ, Lytle CM, Qian JH, Terry N (2001) Removal and distribution of iron, manganese, cobalt and nickel within a Pennsylvania constructed wetland treating coal combustion by-product leachate. *J Environ Qual* 30:1464–1473
- Yoon J, Cao X, Zhou O, Ma LQ (2006) Accumulation of Pb, Cu, and Zn in native plants growing on a contaminated Florida site. *Sci Total Environ* 368:456–464